

Tetrakis(*tert*-butyl 3-oxobutanoato)zirconium(IV)

Usha K. Urs,^a M. S. Dharmaprakash,^b S. A. Shivashankar^b and T. N. Guru Row^{a*}

^aSolid State and Structural Chemistry Unit, Indian Institute of Science, Bangalore 560 012, India, and ^bMaterials Research Centre, Indian Institute of Science, Bangalore 560 012, India

Correspondence e-mail: ssctng@sscu.iisc.ernet.in

Key indicators

Single-crystal X-ray study
T = 293 K
 Mean $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$
R factor = 0.047
wR factor = 0.102
 Data-to-parameter ratio = 19.9

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The structure of the title compound, $[\text{Zr}(\text{C}_8\text{H}_{13}\text{O}_3)_4]$, an MOCVD precursor, has been determined. The zirconium coordination geometry is dodecahedral and the chelate rings are significantly non-planar. The Zr atom lies on a $\bar{4}$ axis, so the asymmetric unit contains only one ligand.

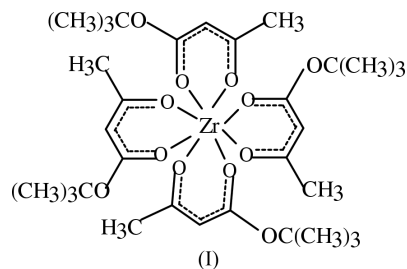
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Comment

The structure of the title zirconium complex, (I), an MOCVD precursor, has been determined. The molecular fourfold inversion symmetry ($\bar{4}$) is retained in the crystal structure, with the Zr atom occupying a special position. The coordination geometry is dodecahedral (Hoard & Silverton, 1963) and not a square antiprism, as in tetrakis(acetylacetonato)-zirconium(IV) (Clegg, 1987; Silverton & Hoard, 1963). The ligand has *tert*-butoxy as a substituent on one side and methyl on the other, as in the Fe complex, also an MOCVD precursor, reported by us previously (Urs *et al.*, 2000). The ligand bite $\text{O}\cdots\text{O}$ distance is 2.703 (2) Å. The six-membered chelate ring is significantly non-planar, the angle between the Zr1/O1/O2 and C1/C2/C3 planes being 3.8 (2)° (Nardelli, 1995). The packing is essentially by van der Waals interactions. There are two intramolecular short contacts of the $\text{C}-\text{H}\cdots\text{O}$ type involving the *tert*-butyl atoms C6 and C7, and the chelate ring atom O2, with $\text{C}\cdots\text{O}$ distances of 2.999 (3) and 2.906 (4) Å, and angles at H of 114.2 (2) and 116.6 (2)°, respectively.



Experimental

The title compound was synthesized by refluxing zirconium *n*-propoxide with *tert*-butyl 3-oxobutanoate in dry benzene, under flowing dry nitrogen. 10 mmol (2.31 g) of zirconium *n*-propoxide was placed in a three-necked round-bottomed flask connected to the dry nitrogen gas line. 40 mmol (6.12 g) of *tert*-butyl 3-oxobutanoate was added, using a pressure equalizer fitted to one of the necks. A reflux condenser fitted with calcium chloride guard tube was fitted to the third neck, to ensure that the system was moisture-free. The reaction mixture was refluxed for 3 h. The propanol-benzene mixture and the excess solvent were removed by distillation. The residue (4.42 g, 85%) was recrystallized from hot *n*-hexane.

Crystal data

[Zr(C₈H₁₃O₃)₄]
M_r = 719.96
 Tetragonal, *I*4₁/a
a = 18.569 (4) Å
c = 10.818 (3) Å
V = 3730.2 (16) Å³
Z = 4
D_x = 1.282 Mg m⁻³

Mo *K*α radiation
 Cell parameters from 19 762 reflections
 $\theta = 2.2\text{--}27.5^\circ$
 $\mu = 0.35\text{ mm}^{-1}$
T = 293 (2) K
 Prism, colourless
 0.19 × 0.13 × 0.05 mm

Data collection

Bruker SMART CCD area-detector diffractometer
 ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
T_{min} = 0.84, *T_{max}* = 0.98
 19 762 measured reflections

2111 independent reflections
 1491 reflections with *I* > 2σ(*I*)
R_{int} = 0.122
 $\theta_{\text{max}} = 27.5^\circ$
h = -23 → 24
k = -23 → 22
l = -14 → 13

Refinement

Refinement on *F*²
R[*F*² > 2σ(*F*²)] = 0.047
wR(*F*²) = 0.102
S = 1.00
 2111 reflections
 106 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0535P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.44\text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.20\text{ e \AA}^{-3}$

Table 1
 Selected geometric parameters (Å, °).

Zr1—O1	2.1008 (16)	C2—C1	1.417 (3)
Zr1—O2	2.2786 (16)	O1—C3	1.283 (3)
C2—C3	1.357 (3)	O2—C1	1.234 (3)
O1 ⁱ —Zr1—O2	71.36 (6)	O1—C3—C2	124.8 (2)
C3—C2—C1	122.3 (2)	O1—C3—C4	115.2 (2)
C3—O1—Zr1	138.39 (15)	C2—C3—C4	120.0 (2)
O2—Zr1—O1—C3	-1.4 (2)	O1—Zr1—O2—C1	7.6 (2)
Zr1—O1—C3—C2	-2.6 (4)	Zr1—O2—C1—C2	-9.9 (3)
C1—C2—C3—O1	2.7 (4)	C3—C2—C1—O2	3.6 (4)

Symmetry codes: (i) $\frac{3}{4} - y, x - \frac{1}{4}, \frac{3}{4} - z$.

H atoms were positioned geometrically and allowed to ride on their respective parent atoms.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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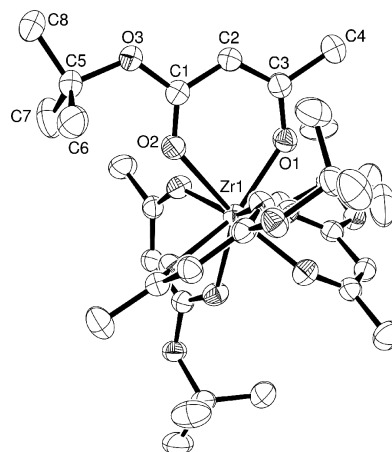


Figure 1

The molecular structure of the title compound, with displacement ellipsoids drawn at the 30% probability level. H atoms have been omitted for clarity and only the atoms of the asymmetric unit are labelled.

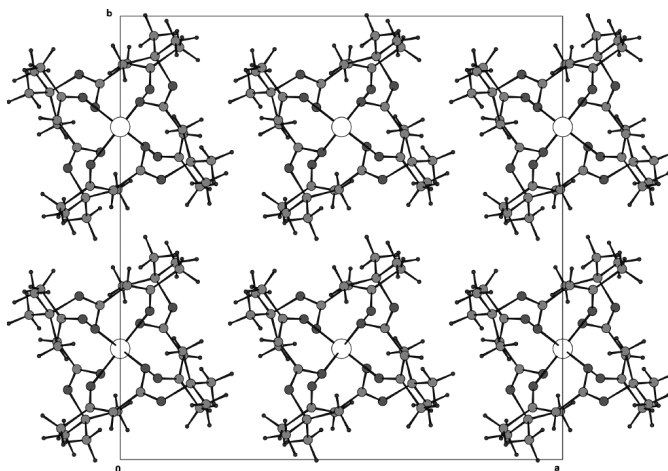


Figure 2

Packing of the molecules, viewed down the *c* axis.

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